Assignment of Absolute Configuration for Virantmycin and Synthesis of Its Antipode

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A total synthesis of antipodal virantmycin was accomplished starting from p-aminobenzoic acid. Absolute stereochemistry of the natural one was shown to be 2S, 3R.

The antibiotic virantmycin( $\underline{1}$ ), isolated by one of us from <u>Streptomyces nitrosporeus</u> in 1981, 1) has been found to possess potent antiviral activity. The planar structure of  $\underline{1}$  has been established by chemical and spectroscopic studies 2) and the synthesis of ( $\underline{+}$ )- $\underline{1}$  was reported by Raphael  $\underline{et}$   $\underline{al}$ . in 1986. 3) However, the relative and absolute stereochemistry at C-2 and C-3 chiral centers has remained to be determined. In this communication we report the determination of the absolute configuration of virantmycin through the synthesis of its (+)-form.

The Wittig reaction of hemi-acetal  $\underline{3}^4$ ) with phosphorane  $\underline{13}^5$ ) afforded (E)-\$\alpha\$,\$\beta\$-unsaturated ester \$\frac{4}{2}\$ (\$\delta\$ 6.57 ppm, 1H, t, J=7 Hz) and its (Z)-isomer (\$\delta\$ 5.60 ppm, 1H, t, J=8 Hz) with the ratio of ca. 30 : 1. The major ester \$\frac{4}{2}\$ was led to allyl alcohol \$\frac{5}{2}\$ by reduction of the ester group and protection of the sulfonamide group. The asymmetric epoxidation of allyl alcohol \$\frac{5}{2}\$ by the usual procedure using (L)-(+)-DET furnished optically active epoxy alcohol \$\frac{6}{2}\$ (90%ee)([\alpha]\_D^{25} -14.9^\circ\$, c 1.00, CHCl\_3) in 98% yield, which was converted to allyl alcohol \$\frac{7}{2}\$ (\$\delta\$ 4.96 ppm, 1H, br s; \$\delta\$ 4.81 ppm, 1H, br s). The alcohol \$\frac{7}{2}\$ was subjected to the metal catalyzed epoxidation (TBHP, VO(acac)\_2/CH\_2Cl\_2/0 °C/2.5 h) to give epoxy alcohol \$\frac{8}{2}\$ as

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a single diastereoisomer in 96% yield. The stereochemistry of 8 was deduced from the configuration of carboxylic ester  $\underline{10}$  (vide  $\underline{infra}$ ). Treatment of the epoxide  $\underline{8}$ with TFA in toluene at room temperature for 6 hours gave piperidine diol 9, which was converted into carboxylic ester  $10^{8}$ ) through protections, deprotections, stepwise oxidation, and monomethylation. The relative stereochemistry of 10 was determined by the <sup>1</sup>H-NMR data and NOE experiments. Observed J values between 3-H and 4-H<sub>2</sub> (J<sub>3-4 $\alpha$ </sub>=6, J<sub>3-4 $\beta$ </sub>=5) suggest an axial orientation of the hydroxyl group at C-3. Relative configuration between C-2 and C-3 substituents was confirmed by the presence of NOE between 17-H $_2$  and 4lpha-H as shown in Fig. 1. In the CD spectrum  $(\Delta \epsilon_{320} = -33.9$ ,  $\Delta \epsilon_{289} = +17.1$ , in EtOH) of 4-dimethylaminobenzoate ester <u>14</u> (4-dimethylaminobenzoic acid, 2-chloro-1-methylpyridinium p-toluenesulfonate,  $^{
m n}$ Bu $_3$ N/ toluene/reflux/4 h), 9) the negative sign of the first Cotton effect was corresponding to the negative chirality. $^{10}$ ) Thus, absolute configuration of 10 was assigned to be 2R, 3S as shown in Fig. 1. The configuration of C-3 to be S was consistent with the expected one from the Sharpless asymmetric epoxidation  $(\underline{5} + \underline{6})$ . Finally, the compound 10 was converted to chloro-ester 11 via an aziridine with double inversions at C-3.

All of spectroscopic data (NMR, IR, MS, UV) and chromatographic behavior of the synthetic chloro-ester  $\underline{11}$  was completely identical with those of the methyl ester  $\underline{2^2}$ ) prepared from natural virantmycin( $\underline{1}$ ). The optical rotation of  $\underline{11}$  was, however, observed as  $[\alpha]_D^{26}$  +8.45° (c 0.250, CHCl $_3$ ) contrary to that of  $\underline{2}$  ( $[\alpha]_D^{26}$  -16.6°, c 0.425, CHCl $_3$ ). Hydrolysis of the chloro-ester  $\underline{11^3}$ ) yielded (+)-virantmycin( $\underline{12}$ ) whose spectra and chromatographic behavior perfectly coincided with natural virantmycin( $\underline{1}$ ). The optical rotation of  $\underline{12}$  ( $[\alpha]_D^{24}$  +11.2°, c 0.125, CHCl $_3$ ) was again different from that of  $\underline{1}$  ( $[\alpha]_D^{24}$  -11.1°, c 0.175, CHCl $_3$ ). The stereochemistry of  $\underline{12}$  was confirmed by NMR data ( $3\alpha$ -H,  $\delta$  4.36, dd, J=6, 5 Hz) and NOE experiments ( $\underline{12a}$  in Fig. 1).

Synthesis of antipodal virantmycin revealed that the absolute configuration of the natural product was shown to be 2S, 3R at the two chiral centers.

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## Reagents and Conditions

**a.**  $\frac{13}{\text{CH}_2\text{Cl}_2/\text{rt}/52}$  h (87%). **b.** 1) DIBAL/toluene/-15 °C/30 min (92%), 2) TMSCI, Et<sub>3</sub>N/CH<sub>2</sub>Cl<sub>2</sub>, 3) AcCl, Et<sub>3</sub>N/CH<sub>2</sub>Cl<sub>2</sub>, 4) citric acid/Et<sub>2</sub>O, MeOH, H<sub>2</sub>O (4:10:1)/rt/50 min (96%, 3 steps). **c.** (L)-(+)-DET, Ti(O<sup>i</sup>Pr)<sub>4</sub>, TBHP/CH<sub>2</sub>Cl<sub>2</sub>/-20 °C/1.5 h (98%). **d.** 1) MsCl, Et<sub>3</sub>N/CH<sub>2</sub>Cl<sub>2</sub>/0 °C/30 min (95%), 2) Nal(5 equiv.), Zn(2 equiv.)/DMF/100 °C/15 min (87%), 3) DIBAL/toluene/-15 °C/30 min (98%). **e.** TBHP, VO(acac)<sub>2</sub>/CH<sub>2</sub>Cl<sub>2</sub>/0 °C/2.5 h (96%). **f.** TFA(2 equiv.)/toluene/rt/6 h (67%). **g.** 1) Me<sub>2</sub>C(OMe)<sub>2</sub>, CSA/acetone/rt/3 h (92%), 2)  $^{n}$ Bu<sub>4</sub>NF/THF/rt/15 h (99%), 3) PDC/CH<sub>2</sub>Cl<sub>2</sub>/rt/4 h,  $^{12}$ ) 4) MnO<sub>2</sub>, KCN, AcOH/MeOH, benzene (2:1)/rt/13 h,  $^{13}$ ) 5) KOH/MeOH, Et<sub>2</sub>O, H<sub>2</sub>O (2:2:1)/rt/24 h (quant., 3 steps), 6) Na, naphthalene/DME/-15 °C/30 min (quant.), 7) CH<sub>2</sub>N<sub>2</sub>/MeOH/rt, 8)  $_{P}$ -TsOH/MeOH, Et<sub>2</sub>O (4:1)/rt/18 h (92%, 2 steps), 9) NaH,  $^{n}$ Bu<sub>4</sub>NI/THF/0 °C/30 min, then MeI(5 equiv.), HMPA(2 equiv.)/-15 °C/1.5 h (70%). **h.** 1) Ph<sub>3</sub>P, DEAD/THF/rt/3 h (89%),  $^{14}$ ) 2) Et<sub>4</sub>NCI (20 equiv.), TFA/CH<sub>2</sub>Cl<sub>2</sub>/-15 °C/30 min (84%). **i.** LiOH/aq.CH<sub>3</sub>CN (50%).  $^{3}$ 

Scheme 1.

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- 5) Phosphorane 13 was prepared from 1-bromo-2,3-dimethyl-2-butene(16) by the following sequence of reactions; 1) NaCH(CO<sub>2</sub>Me)<sub>2</sub>/THF (86%), 2) NaCl, H<sub>2</sub>O/DMSO, 3) LAH, 4) MsCl, Et<sub>3</sub>N/CH<sub>2</sub>Cl<sub>2</sub> (86%, 3 steps), 5) NaI/DMF, 6) Ph<sub>3</sub>P/benzene/reflux (89%, 2 steps), 7) LiN(TMS)<sub>2</sub>/THF/0 °C/30 min, then ClCO<sub>2</sub>Me/-78 °C/2 h (88%). For the preparation of 16, see; L. Ruzicka and H. Schinz, Helv. Chim. Acta, 23, 959 (1940).
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